

CODEX ALIMENTARIUS COMMISSION



Food and Agriculture
Organization of
the United Nations



World Health
Organization

Viale delle Terme di Caracalla, 00153 Rome, Italy - Tel: (+39) 06 57051 - Fax: (+39) 06 5705 4593 - E-mail: codex@fao.org - www.codexalimentarius.net

Agenda Item 4

CX/CF 11/5/4
November 2010

JOINT FAO/WHO FOOD STANDARDS PROGRAMME

CODEX COMMITTEE ON CONTAMINANTS IN FOODS

5th Session

The Hague, The Netherlands, 21 – 25 March 2011

PROPOSED DRAFT CODE OF PRACTICE FOR THE PREVENTION AND REDUCTION OF
ETHYL CARBAMATE IN STONE FRUIT SPIRITS

(N11-2009)

(At Step 3)

Prepared by Electronic Working Group led by Germany

Codex Members and Observers wishing to submit comments at Step 3 on the above matter, including possible implications for their economic interests, should do so in conformity with the *Uniform Procedure for the Elaboration of Codex Standards and Related Texts* (Codex Alimentarius Commission Procedural Manual) before **15 January 2011**. Comments should be directed:

to:

Mrs Tanja Åkesson
Codex Contact Point
Ministry of Agriculture, Nature and Food Quality
P.O. Box 20401
2500 EK The Hague
The Netherlands
Fax.: +31 70 378.6134
E-mail: info@codexalimentarius.nl - *preferably* -

with a copy to:

Secretariat, Codex Alimentarius Commission,
Joint FAO/WHO Food Standards Programme,
Viale delle Terme di Caracalla,
00153 Rome, Italy
Fax: +39 (06) 5705 4593
E-mail: codex@fao.org - *preferably* -

BACKGROUND

1. The 3rd Session of the Codex Committee on Contaminants in Foods agreed to start new work on a Code of Practice for the Prevention and Reduction of Ethyl Carbamate in Stone Fruit Distillates which was approved by the 32nd Session of the Commission.¹
2. The 4th Session of the Committee considered the Proposed Draft Code of Practice for the Prevention and Reduction of Ethyl Carbamate in Stone Fruit Distillates and agreed to return the proposed draft Code to Step 3 for further comments.²
3. The Committee further agreed to establish an electronic working group led by Germany that would prepare a revised version of the COP based on the comments submitted at Step 3 for consideration by the next session of the Committee.
4. Comments were received from Brazil, Canada, European Union and FAO and were made available to the EWG.

¹ ALINORM 09/32/REP, para. 113 and App. VI.

² ALINORM 10/33/41, paras 43 – 54.

5. The EWG revised the proposed draft Code of Practice. Apart from certain editorial revisions, the main changes relate to the clarification of measures proposed for removal of hydrocyanic acid and the prevention of its formation during the process of distillation (para. 21 and 22). Furthermore, the term “stone fruit distillate” was specified (para. 7) and applied consistently in the COP.

6. The Proposed Draft Code of Practice as prepared by the electronic working group is hereby circulated for comments at Step 3 (see Appendix I). The list of participants of the electronic working group is attached as Appendix II.

APPENDIX I

PROPOSED DRAFT CODE OF PRACTICE FOR THE PREVENTION AND REDUCTION OF
ETHYL CARBAMATE CONTAMINATION IN STONE FRUIT DISTILLATES

(AT STEP 3 OF THE PROCEDURE)

INTRODUCTION

1. Ethyl carbamate is a compound that occurs naturally in fermented foods and alcoholic beverages such as bread, yoghurt, soy sauce, wine, beer, and particularly in stone fruit spirits and stone fruit marc spirits, mainly those made from cherries, plums, mirabelles and apricots.
2. Ethyl carbamate can be formed from various substances inherent in food and beverages, including hydrogen cyanide (or hydrocyanic acid), urea, citrulline, and other N-carbamyl compounds. Cyanate is probably the ultimate precursor in most cases, reacting with ethanol to form ethyl carbamate. Therefore ethyl carbamate reduction measures should focus on hydrocyanic acid and other precursors of ethyl carbamate.
3. Ethyl carbamate is genotoxic and a multisite carcinogen in animals and is probably carcinogenic to humans.
4. Stone fruit and stone fruit marc spirits, in particular, contain ethyl carbamate in manyfold higher concentrations than other fermented foods and beverages. In stone fruit distillates (such as stone fruit spirits and stone fruit marc spirits) ethyl carbamate can be formed from cyanogenic glycosides that are natural constituents of the stones. When mashing the fruit, the stones may be damaged and cyanogenic glycosides from the stones may come into contact with enzymes in the fruit mash. Cyanogenic glycosides are then degraded to hydrocyanic acid/cyanides. Hydrocyanic acid may also be released from intact stones during a prolonged storage of the fermented mash. During the distillation process hydrocyanic acid may be enriched in all fractions. Cyanide in the distillates may be oxidized to cyanate, which can react with ethanol to form ethyl carbamate. Certain environmental conditions such as exposure to light, high temperatures and the presence of copper ions promote the formation of ethyl carbamate in the distillate.
5. Although no strong correlation between the level of hydrocyanic acid and ethyl carbamate has been established so far, it is evident that under certain conditions high concentrations of hydrocyanic acid lead to higher levels of ethyl carbamate. A potential increase in ethyl carbamate formation has been associated with levels at or above 1 mg/l hydrocyanic acid in the final distillate. Based on practical experiences it can be assumed that from 1 mg of hydrocyanic acid up to 0.4 mg ethyl carbamate potentially can be formed in a non-equimolar relationship.

SCOPE AND DEFINITIONS

6. This Code of Practice intends to provide national and local authorities, manufacturers and other relevant bodies with guidance to prevent and/or reduce formation of ethyl carbamate in stone fruit spirits and stone fruit marc spirits. Ethyl carbamate formation in other alcoholic beverages and foods is not covered in this Code.
7. The definitions below apply to this Code:
 - (a) **Stone Fruit** means a fruit which is produced on trees belonging to the genus *Prunus* of the rose family (Rosaceae).
 - (b) **Distillates** means, for the purpose of this Code of Practice, alcohol-rich products obtained after the distillation process and ready for consumption.
 - (c) **Stone Fruit distillates** means, for the purpose of this Code of Practice, the distillates for consumption, obtained after the distillation:
 - of the mash prepared by fermentation of crushed stone fruits
 - of fermented stone fruit marc (pomace).
 - of mash obtained by fermentation and/or maceration of crushed and/or whole stone fruit in ethyl alcohol or alcoholic beverages.

GENERAL REMARKS

8. This Code covers all possible measures that have been proven to prevent and/or reduce high levels of ethyl carbamate in stone fruit distillates. When applying the Code for specific stone fruit distillates, measures should be

carefully chosen from the viewpoint of benefit and feasibility. In addition, measures should be implemented in accordance with the relevant national and international legislation and standards.

9. It is recognised that reasonably applicable technological measures - Good Manufacturing Practices (GMP) - can be taken to prevent and reduce significantly high ethyl carbamate levels in stone fruit distillates. The reduction of ethyl carbamate could be achieved using two different approaches: first, by reducing the concentration of the main precursor substances (e.g. hydrocyanic acid and cyanides); second, by reducing the tendency of these substances to react to form cyanate.

TYPICAL PRODUCTION PROCESS

10. The production process for stone fruit distillates involves preparing mash by using whole stone fruits or their marc as ingredients, followed by fermentation and distillation. The process typically follows the steps listed below:

- (a) Preparing mash by crushing the whole ripe fruit for stone fruit spirit drinks or by using stone fruit marc for stone fruit marc spirit drinks;
- (b) fermenting the mash in stainless steel tanks or other suitable fermentation vessels;
- (c) in the case of using a maceration process, the mash is prepared by macerating crushed or whole fruit in ethyl alcohol or alcoholic beverages and stored for a period, without fermentation process;
- (d) transferring the fermented mash into the distillation device, often a copper pot;
- (e) heating the fermented mash by a suitable heating method in order to slowly boil off the alcohol;
- (f) cooling the alcohol vapour in an appropriate (e. g. stainless steel) column where it condenses and is collected;
- (g) separation of three different fractions of alcohol: 'heads', 'hearts' and 'tails';
- (h) dilution to the final alcoholic grade.

11. During distillation, the heads boil off first. Components with low boiling point e.g. ethyl acetate and methanol are part of the heads. This fraction is generally unsuitable for consumption and should be discarded.

12. During the middle distillation run (the 'hearts'), the principal alcohol in all spirits, ethyl alcohol (ethanol), is distilled. This part of the distilling run, where the content of volatiles other than ethanol is lowest and the purest fruit aromas are present, is always collected.

13. The 'tails' of the distillation include acetic acid and fusel oils, which are often identified by unpleasant vinegary and vegetal aromas. They are also discarded, but they may be re-distilled because some ethanol is invariably included with the tails.

RECOMMENDED PRACTICES BASED ON GMP's

RAW MATERIALS AND PREPARATION OF FRUIT MASH

14. The raw materials and preparation of the fruit mash should be suitable to avoid the release of hydrocyanic acid, a precursor of ethyl carbamate.

15. The stone fruits should generally be of a high quality, not mechanically damaged and not microbiologically spoiled, as damaged and spoiled fruit may contain more free cyanide.

16. The fruit should preferably be de-stoned.

17. If the fruits are not de-stoned and/or the residues of fruits (marc) are used for preparing mash, they should be mashed gently avoiding the crushing of stones. If possible, stones should be removed from the mash.

FERMENTATION

18. Selected yeast preparations for the production of spirit drinks should be added to the mashed fruits, according to the manufacture's instructions for users, for a fast and "clean" fermentation.

19. Mashed fermented fruits should be handled with high standards of hygiene, and exposure to light should be minimised. Fermented fruit mashes containing stones should be stored as briefly as possible before distillation since hydrocyanic acid may also be released from intact stones during prolonged storage.

20. If the mash is prepared by macerating stone fruit into alcoholic beverages, the stone fruit should be removed soon after the aroma of stone fruit is adequately extracted.

DISTILLATION EQUIPMENT

21. Distillation equipment and the distillation process should be suitable, to ensure that hydrocyanic acid is not transferred into the distillate

- (a) Use of a copper still will limit carryover of ethyl carbamate-forming precursors into the distillate.
- (b) The distillation equipment should preferably include automatic rinsing devices and copper catalytic converters. The automatic rinsing devices will keep the copper stills cleaned while the copper catalytic converters will bind hydrocyanic acid before it passes into the distillate.
- (c) Automatic rinsing devices are not necessary in the case of discontinuous distillation. The distillation equipment should be cleaned by systematic and thorough cleaning procedures.
- (d) When copper catalytic converters or other dedicated cyanide separators are not available, copper (I) chloride preparations can be added to the fermented fruit mash before distillation. The purpose of these preparations containing copper (I) ions is to bind hydrocyanic acid before it passes into the distillate. Copper (II) ions are without effect and should not be used.

22. . While copper ions can inhibit formation of ethyl carbamate precursors in the mash and in the still, they can promote formation of ethyl carbamate in the distillate. Therefore, use of a stainless steel condenser at the end of the distillation device rather than a copper condenser will limit presence of copper in the distillate and reduce the rate of ethyl carbamate formation.

DISTILLATION PROCESS

23. Stones settled in the fermented mash should not be pumped into the distillation device.

24. Distillation should be carried out in such a way that alcohol is boiled off slowly and in a controlled matter (e.g. by using steam instead of a direct flame as the heating source).

25. The first fractions of the distillate, called 'heads', should be separated carefully.

26. The middle fraction, called 'hearts', should then be collected and should be stored in the dark. When the alcohol content of the actual distillate reaches 50% vol. at the receiver, collection should be switched to the 'tails', so that any ethyl carbamate that may have been formed is separated in the tail fraction.

27. Some manufacturers may redistill the separated tails, possibly containing ethyl carbamate. If the tails are used for re-distilling, they should be re-distilled separately, however for reduction of ethyl carbamate concentration it is preferable to discard the tail.

CHECKS ON THE DISTILLATE, RE-DISTILLATION AND STORAGE

Hydrocyanic acid

28. Testing for hydrocyanic acid may be used as a simple test for ethyl carbamate in distillates. Therefore, the distillates should be regularly checked for their levels of hydrocyanic acid. The determination could be carried out by specific tests including kits for rapid testing of the hydrocyanic acid levels.

29. If the concentration of hydrocyanic acid in the distillate exceeds a level of 1 mg/l, re-distillation with catalytic converters or copper preparations is recommended (see point 21).

30. Distillates should be stored in bottles that are lightproof (or filter ultraviolet light) or in covering boxes and not at higher temperatures.

Ethyl carbamate

31. Testing of ethyl carbamate is recommended for distillates in which the compound may already have been formed (e.g. distillates with unknown history of production, distillates with higher levels of hydrocyanic acid, or storage at light or at high temperatures).

32. Additional distillation is effective in order to reduce ethyl carbamate in distillates (see point 26).

GENERAL RECOMMENDATIONS

33. The national, state and local governments as well as the non-governmental organizations (NGOs, commercial associations and cooperatives) should provide their own basic training and update the information on mitigating ethyl carbamate in stone fruit spirits and fruit marc spirits.

34. The non-industrial, small-scale preparation of these drinks should have available material with information on the specific recommendations based on good manufacturing practices and guidance on prevention and reduction of ethyl carbamate in the stone fruit distillates. Specifically, material should be made available to small-scale producers of stone fruit spirits.

Appendix II

Federal Public Service Health, Food Chain Safety and Environment
 DG Animal, Plant and Food
 Service Foodstuffs, Feed and Other Products
Isabel.deboosere@health.fgov.be
 Tel + 32 2 524 73 84
 Fax + 32 2 524 73 99
 Place Victor Hortaplein 40 box 10
 1060 Brussels
 Belgium

Ms. LÍgia Lindner Schreiner
 Expert on Regulation
 Brazilian Health Surveillance Agency
 General Office of Foods
 Tel.: +55 61 3462 5399
 E-mail: ligia.schreiner@anvisa.gov.br and
gacta@anvisa.gov.br

Luc Pelletier
 Chemical Health Hazard Assessment Division
 Bureau of Chemical Safety
 Food Directorate
 Health Products and Food Branch, Health Canada
 1st Floor East, AL: 2201C
 Building 22, Tunney's Pasture
 Ottawa, ON Canada
 K1A 0K9
 Telephone: 613-946-9089
 Email: luc.pelletier@hc-sc.gc.ca

The European Union
 Ms Almut Bitterhof: almut.bitterhof@ec.europa.eu
 and codex@ec.europa.eu

Mrs. Ágnes Palotásné Gyöngyösi
 Ministry of Rural Development, Hungary
 e-mail: Agnes.Gyongyosi@fvm.gov.hu

Dr Fumi IRIE
 Deputy Director
 Standards and Evaluation Division, Department of Food
 Safety, Ministry of Health, Labour and Welfare
 Address: 1-2-2 Kasumigaseki, Chiyoda-ku, Tokyo 100-8916, Japan
 Phone: +81-3-3595-2341
 Fax: +81-3-3501-4868
 E-mail: codexj@mhlw.go.jp

Mr Takashi IJIMA
 Technical Officer (Analysis and Brewing Technology)
 Taxation Department, National Tax Agency
 E-mail: Codex@nta.go.jp, takashi.ijima@nta.go.jp
 Telephone: +81-3-3581-0180
 FAX: +81-3-3581-4747

Dr Yoshihiko OE
 Technical Officer (Analysis and Brewing Technology)
 Technical Advisory office, Second Taxation
 Department,
 Tokyo Regional Taxation Bureau
 E-mail: yoshihiko.oe@tok.nta.go.jp
 Telephone: +81-3-3910-6235
 FAX: +81-3-3910-3398

Mr Tomokazu HASHIGUCHI
 Senior Researcher
 Safety and Quality Research Division, National
 Research
 Institute of Brewing, Ind. Adm.
 E-mail: hashiguchi@nrib.go.jp
 Telephone: +81-82-420-0800
 FAX: +81-82-420-0804

Ana Biel Canedo
 Ana López-Santacruz Serraller
 Subdirector-General for Food Risk Management
 Spanish Food Safety and Nutrition Agency
contaminantes@msps.es

Lauren Posnick Robin, Sc.D.
 Review Chemist
 Center for Food Safety and Applied Nutrition
 U.S. Food and Drug Administration
 HFS-317
 5100 Paint Branch Parkway
 College Park, MD 20740
 301-436-1639 (Phone)
 301-436-2651 (Fax)
lauren.robin@fda.hhs.gov

Dr Annika Wennberg
 FAO JECFA Secretary
 Nutrition and Consumer Protection Division
 Food and Agriculture Organization of the United
 Nations
 Viale delle Terme di Caracalla, C- 278
 00153 Rome, Italy
 Telephone: + 39 06 5705 3283
 Facsimile: + 39 06 5705 4593
 E-mail: Annika.Wennberg@fao.org

Lorcan O' Flaherty
 Affiliation: Confederation of the Food and Drink
 Industries of the EU (CIAA)
 Telephone: +32 2 5008756
 Fax: +32 2 5112905
 Email: l.oflaherty@ciaa.eu