

## POLYOXYETHYLENE (20) SORBITAN MONOPALMITATE

*Prepared at the 17th JECFA (1973), published in FNP 4 (1978) and in FNP 52 (1992). Metals and arsenic specifications revised at the 55th JECFA (2000). An ADI of 0-25 mg/kg bw was established at the 17th JECFA (1973)*

### SYNONYMS

Polysorbate 40; INS No. 434

### DEFINITION

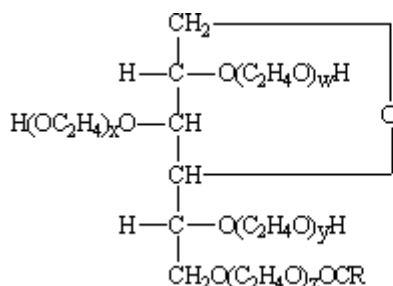
Consists of a mixture of the partial esters of sorbitol and its mono- and dianhydrides (which have an acid value below 7.5 and a water content below 0.2%) with edible commercial palmitic acid condensed with approximately 20 moles of ethylene oxide per mole of sorbitol and its anhydrides.

C.A.S. number

9005-66-7

Structural formula

Nominal formula and approximate composition:



where  $w + x + y + z = \text{approx. } 20$  and  $\text{RCO-}$  is the fatty acid moiety

Assay

Not less than 66.0 and not more than 70.5% of oxyethylene groups, equivalent to not less than 97.0 and not more than 103.0% of polyoxyethylene (20) sorbitan monopalmitate calculated on the anhydrous basis.

### DESCRIPTION

Lemon to orange coloured, oily liquid or semi-gel at 25°, with a faint characteristic odour

**FUNCTIONAL USES** Emulsifier, dispersing agent

### CHARACTERISTICS

#### IDENTIFICATION

Solubility (Vol. 4)

Soluble in water, ethanol, methanol, ethyl acetate and acetone; insoluble in mineral oil

Infrared absorption

The infrared spectrum of the sample is characteristic of a partial fatty acid ester of a polyoxyethylated polyol

Colour reaction

To 5 ml of a 5% (w/v) aqueous solution of the sample add 10 ml of ammonium cobalthiocyanate solution and 5 ml of chloroform, shake well

and allow to separate; a blue colour is produced in the chloroform layer.  
(Ammonium cobalthiocyanate solution: 37.5 g of cobalt nitrate and 150 g of ammonium thiocyanate made up to 100 ml with water - freshly prepared).

Test for fatty acids

To 5 ml of a 5% (w/v) aqueous solution of the sample add 5 ml sodium hydroxide TS. Boil for a few min, cool, and acidify with dilute hydrochloric acid. The solution is strongly opalescent, owing to the fatty acids liberated.

Gelatinization

A mixture of 60 parts by volume of the sample and 40 parts of water yields a gelatinous mass at or below room temperature

Saponification (Vol. 4)

100 g of the sample yields approximately 20 g of fatty acids and 78 g of polyols

PURITY

Water (Vol. 4)

Not more than 3% (Karl Fischer Method)

Sulfated ash (Vol. 4)

Not more than 0.25%  
Test 5 g of the sample

Acid value (Vol. 4)

Not more than 2

Saponification value  
(Vol. 4)

Not less than 41 and not more than 52

Hydroxyl value (Vol. 4)

Not less than 90 and not more than 107

Lead (Vol. 4)

Not more than 2 mg/kg  
Determine using an atomic absorption technique appropriate to the specified level. The selection of sample size and method of sample preparation may be based on the principles of the method described in Volume 4, "Instrumental Methods."

**METHOD OF ASSAY**

Determine the content of *Oxyethylene groups*.