

Monitoring C storage and loss through isotope technology

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INTRODUCTION

What are the benefits of using stable isotope technology in **SOC studies?**

SOC studies are often labortime-consuming intensive, and limited to assessing total organic C stocks. The role of various carbon compounds and crops in SOC storage and loss can be identified with stable isotope technology.

OBJECTIVES

What work is being **conducted in the SWMCNL?**

The Soil and Water Management & Crop Nutrition Laboratory is:

> Developing methods of calibration data and

MAIN RESULTS

Progress to date?

Reference gases mixed with synthetic air (Fig. 1) were determined to be appropriate for ¹³C-CO₂laser stable isotope spectroscopy (Fig. 2).

Instrumentation calibration and





What does stable isotope technology measure?

Stable technology isotope differentiates sources of SOC storage and loss by measuring differences in the natural (12C)abundance of carbon-12 carbon-13 (13C) in plant and soil carbon pools and using artificially labelled 13C tracers. Accurate estimations of plant C contributions to soils are only possible with homogenous ¹³C-labelling of plant material. Labelled plants are increasingly used to trace the fate of plant-derived C in soils. Classic laboratory instrumentation measures amounts of carbon and the proportion of ¹²C and ¹³C isotopes in a sample. Laser stable isotope spectroscopy is an emerging technology that will provide more robust data on sources of CO₂ lost from soils. The instrumentation can be transported to a desired location to take in situ measurements and can take continuous, real-time measurements without delays in sample collection and processing.

correction for laser stable for isotope spectroscopy accurate and precise data;

- Testing analysis of sources of soil C loss with laser stable isotope spectroscopy in a soil mulch-application study;
- Increasing resolution of soil C storage and loss analysis in the mulch-application study using classic methods of SOC analysis;
- ➢ Producing homogenously ¹³Clabelled plant material as well as pulse ¹³C-labelled plant material to trace storage and loss of plant-derived C in soils.

data correction to be validated;

Measurements of fluxes of ¹³C-CO₂ from soils has been achieved (Fig. 3).

evaluating Processing and methods for large amounts of CO₂ and $\delta^{13}C$ data have been established;

- Effects of superficial mulchapplication on soil C and N analyzed 4 yr after experimental setup (Table 1). Mulch effects mostly observed in top soil. Microbial C slightly responsive to mulch, SOC content is not. SOC $\delta^{13}C$ suggest a shift in C contributing processes. More mulch effects in high OM soils;
- > High throughput homogenous ¹³C-labelling of maize has been achieved (Fig. 4 & Table 2).

Fig. 3: Experimental meso-cosms. Agricultural soil being sampled for CO₂. For 4 ys, maize and soybean have been grown in rotation and no mulch/ mulch has been applied as an experimental treatment. Water has been maintained constant in treatments



Fig. 4: Maize plants grown in growth chamber with ¹³CO₂ labelling. Scrubbers are suspended to reduce ethylene levels and a desiccant rests on the floor to reduce humidity. Fans promote structural stability of maize

CONCLUSION



Fig. 2: ¹³C-CO₂ laser stable isotope analyzer for measuring CO₂ emissions and δ^{13} C of CO₂ from soils

soybean has also been achieved. Dual-labelled plants will be produced this year.

Tab. 2: δ^{13} C of stabile components (lignin and cellulose) and labile components (proteins, carbohydrates and organic acids) of plant material collected along the length of maize

	Leaf	Intra-leaf	δ ¹³ C stabile	δ ¹³ C labile
\mathcal{N}	top	stem	399	404
\sim	top	tip	398	403
	middle	stem	398	407
	middle	tip	378	395
	bottom	stem	396	406
	bottom	tip	282	305

Tab. 1: C and N analysis of soil organic C and N as well as microbial C and N of meso-cosm soils (Fig. 1) with low (L) or high (H) organic matter without mulch (OM) and with mulch (M). Each value represents the mean of 3 replicates and standard errors. Significant differences in values due to organic matter and mulch are shown for P < 0.05

0-5 cm	LOM	LM	HOM	HM
SOC δ^{13} C	-26.24 ± 0.03^{ac}	-25.69 ± 0.23 ^{bd}	-26.27 ± 0.06 ^{ab}	-25.76 ± 0.06 ^{cd}
Soil mg organic C g ⁻¹ soil	10.82 ± 0.11^{a}	12.56 ± 0.58^{a}	24.66 ± 1.55^{b}	26.39 ± 0.54 ^b
Microbial mg C g ⁻¹ soil	0.011 ± 0.006^{abc}	$0.063 \pm 0.004^{\text{abc}}$	0.054 ± 0.031^{b}	0.154 ± 0.023 ^c
δ^{15} N	7.54 ± 0.22^{a}	8.00 ± 0.16^{a}	8.38 ± 0.29^{b}	$8.60 \pm 0.30^{\rm b}$

Additional ¹⁵N-labelling of **Stable isotope techniques for** climate-smart agriculture

At the joint FAO/IAEA SWMCN laboratory isotope techniques are being established and combined to evaluate practices for climate-smart agriculture. Specifically, mulch applications are being evaluated for the capacity to:

- \succ Reduce soil CO₂ emissions due to the large potential for soils to emit this greenhouse gas to the atmosphere;
- Store SOC belowground to improve soil quality and reduce atmospheric levels of CO₂.

Mulch applications are also being evaluated for...

> Managing fertilizer application to improve plant yield and



Fig. 1: Gas mixing line linked to a vacuum pump, pressure gauge, CO_2 and synthetic air gas tanks to make reference gases for laser

stable isotope analyzers

mg N g ⁻¹ soil Microbial mg N g ⁻¹ soil	$1.38 \pm 0.04^{\text{abc}}$	$1.43 \pm 0.05^{\text{abc}}$	2.26 ± 0.04^{b}	2.55 ± 0.09 ^c 0.026 ± 0.005 ^b
Microbial mg N g ⁻¹ soil	0.015 ± 0.005^{a}	0.015 ± 0.002^{a}	$0.028 \pm 0.002^{\rm b}$	$0.026 \pm 0.005^{\mathrm{b}}$

5-15 cm	LOM	LM	НОМ	HM
SOC δ^{13} C	$-26.53 \pm 0.04^{\text{abc}}$	$-26.42 \pm 0.05^{\text{abc}}$	-26.28 ± 0.09^{b}	-26.11 ± 0.05°
Soil mg organic C g ⁻¹ soil	10.66 ± 0.39^{a}	10.34 ± 0.05^{a}	23.89 ± 0.50 ^b	24.00 ± 0.52^{b}
Microbial mg C g ⁻¹ soil	0.041 ± 0.011^{a}	0.040 ± 0.004^{a}	0.177 ± 0.028^{b}	0.147 ± 0.022^{b}
$\delta^{15}N$	7.07 ± 0.14^{a}	7.25 ± 0.12^{a}	8.25 ± 0.10^{b}	7.99 ± 0.21 ^b
mg N g ⁻¹ soil	1.27 ± 0.02^{a}	1.23 ± 0.02^{a}	2.29 ± 0.05 ^b	2.29 ± 0.05 ^b
Microbial mg N g ⁻¹ soil	0.016 ± 0.0003 ^a	0.018 ± 0.008^{a}	0.021 ± 0.004 ^a	0.022 ± 0.004^{a}

reduce pollution;

 \succ Reducing soil N₂O emissions due to the large potential for release fertilizers to this greenhouse gas.

Stable isotope techniques will be combined into guidelines for Member States to use for climatesmart agriculture.

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