

### **Hexazinone (DPX-A3674)**

## **Determination of Ethyl Carbamate (IN-08387) in Hexazinone Technical Headspace Gas Chromatographic (GC) Trace Level Method**

### **I. Principle**

The sample is heated in the headspace unit until equilibrium is reached between the solid and the vapor phases. The vapor containing ethyl carbamate (ETC), as well as the other volatile components, is automatically injected into the gas chromatograph. Separation of ethyl carbamate from sample components is accomplished on a J&W DB-Wax column and detected using a flame ionization detector. External standard technique is used for method calibration. The weight percent of ethyl carbamate in each sample is determined by comparison to a calibration curve (peak area vs. concentration) prepared from the analysis of standards prepared by spiking hexazinone technical with ethyl carbamate.

### **II. Applicability**

This method is applicable for the determination of ethyl carbamate in technical hexazinone samples at concentrations ranging from 10 ppm to 50 ppm.

This method was developed to include validation data on linearity, precision (repeatability/reproducibility), accuracy (recovery), limitations and interferences. The data was generated in accordance with U.S. EPA Good Laboratory Practice Standards (40 CFR Part 160), which are consistent with the OECD Principles of Good Laboratory Practice. (Comment 1)

### **III. Limitations**

There are no known chromatographic interferences with ethyl carbamate employing the conditions specified in the method.

Headspace analysis relies on equilibration between the solid, liquid and vapor phase in both the sample and the standard. The standard and sample should have as close to the same weight and chemical make-up as possible.

#### IV. Statistical Characterization

##### A. Linearity

Linearity of the calibration curve (area vs. amount) and sensitivity was demonstrated by analysis of preparations of analytical standards at five different levels. The method of linear least squares was used to determine the equation for the straight line that best fits the experimental data. The following values were obtained for the slope, y-intercept, and correlation coefficient:

Amount range (ethyl carbamate): = 10 - 50  $\mu\text{g/g}$

Slope (sensitivity): = 40.751 (area /amount)

Intercept: = 65.284 (area)

Correlation Coefficient (r): = 0.9988

##### B. Precision

###### 1. Repeatability

The repeatability of the method for ethyl carbamate was determined utilizing 10 replicate test portions of one sample (DPX-A3674-341) spiked with ethyl carbamate at 25 ppm, analyzed by one analyst on the same day.

The mean range ( $\bar{X}$ ), standard deviation ( $\sigma$ ) and relative standard deviation (RSD) for analyses are as follows:

Component	$\bar{X}$ Range	$\sigma$	RSD
ethyl carbamate, ppm	26.7	1.5	5.6%

There was 1 outlier in the data set which was removed for statistical analysis.

###### 2. Reproducibility

The reproducibility of the method for ethyl carbamate was determined utilizing 10 replicate test portions of one sample (DPX-A3674-341) spiked with ethyl carbamate at 25 ppm, analyzed in 2 labs on two days. (Comment 1)

The mean range ( $\bar{X}$ ), standard deviation ( $\sigma$ ) and relative standard deviation (RSD) for analyses are as follows:

Component	$\bar{X}$ Range	$\sigma$	RSD
ethyl carbamate, ppm	28.3	4.6	16.3%

There was 1 outlier in the data set from day 1 which was removed for statistical analysis.

C. Accuracy

Duplicate test portions of hexazinone technical (DPX-A3674-341) were spiked at two levels (nominally 25 ppm and 35 ppm) with a solution containing a predetermined amount of ethyl carbamate. Each sample was then analyzed according to the sample procedure specified in this method. The average recoveries are as follows:

ppm Spiked	ppm Recovered	% Recovery
23.7	21.7	91.7%
24.6	24.6	99.9%
32.6	36.4	111.6%
33.7	36.5	108.3%

The average recovery for the analysis was 102.9%.

D. Limits of Detection and Quantitation

The limit of quantitation of the method for ethyl carbamate was demonstrated utilizing 5 replicate test portions of one sample (DPX-A3674-341) spiked with ethyl carbamate at 10 ppm, analyzed by one analyst on the same day.

The mean range ( $\bar{X}$ ), standard deviation ( $\sigma$ ) and relative standard deviation (RSD) for analyses are as follows:

Component	$\bar{X}$ Range	$\sigma$	RSD
ethyl carbamate, ppm	10.8	1.3	11.9%

The limit of quantitation for the method has been demonstrated to be at least 10 ppm.

**V. Special Apparatus (Equivalent apparatus may be substituted)**

1. Gas chromatograph: HP 6890 Series GC, equipped with split/splitless injector, a flame ionization detector and Electron Pressure Control. Hewlett-Packard Company (currently Agilent Technologies), 2850 Centerville Road, Wilmington, DE 19808-1610.
2. Headspace Sampler: HP 7694 equipped with auto-sampler trays and heating oven, Hewlett-Packard Company (currently Agilent Technologies), 2850 Centerville Road, Wilmington, DE 19808-1610. (Comment 2)
3. Analytical column: 30 m x 0.53 mm i.d., 1- $\mu$ m film thickness, DB Wax, J & W Scientific, Catalog No. 1257032, (J & W Scientific, Folsom, CA 95630).
4. Sample vials: septa and caps, 20 mL, Hewlett Packard, Catalog No. 19395-280020, (Hewlett Packard Co., Valley Forge, PA 19842).
5. Injection liner: split/splitless inlet sleeve for HP 6890 Series GC, SGE FocusLiner manufactured for Agilent, 78.5 X 6.3 X 4 mm, Part No. 210-4004-5, supplied by VWR, Catalog No. 21513-680.
6. Micropipettor: Eppendorf Electronic Pipettor, 4850 Series, 100- $\mu$ L variable volume, VWR, 53512-207.

**VI. Reagents (Reagent grade except as noted)**

1. n-Hexane, Fischer Scientific Catalog No. H3021, (Fischer Scientific, Houston, TX 77099).
2. Hexazinone, reference standard grade, (DuPont Crop Protection, Stine-Haskell Research Center, Newark, DE 19714-0030).
3. Ethyl carbamate (urethane), Aldrich Chemical Co. Catalog No. U285-7, (Aldrich Chemical Co. Milwaukee, WI 53201).
4. Nitrogen, Zero grade.

**VII. Special Safety Considerations**

**A. Product Hazards**

Hexazinone is corrosive. Exposure to hexazinone by eye contact may cause eye irritation with discomfort, tearing, or blurring of vision. Ingestion may result in abnormal liver function. Avoid contact with eyes and skin. Wear gloves when handling. Wash thoroughly after handling. In case of skin or eye contact, flush with copious amounts of water. Contact a physician. Consult MSDS #M0000071 for detailed information.

**B. Procedure Hazards**

Ethyl carbamate is a cancer suspect agent. Do not breathe ETC dust. Avoid contact with eyes, skin and clothing. Refer to area MSDS #B0001381 for detailed information.

n-Hexane is a flammable liquid. Avoid skin contact. Refer to area MSDS #L0000758 for detailed information.

**VIII. Procedure**

**A. Operating Conditions**

Chromatographic Conditions:

Gas chromatograph	
Type:	HP
Model Number:	6890
Carrier gas:	
Type:	He (Comment 3)
Mode :	Constant flow
Flow rate, mL/min:	6
Injection port:	
Temperature, °C:	250
Injection mode:	Splitless
Purge, min:	0.75
Purge flow, mL/min:	15
Column:	
Length, m:	30
I. D., mm:	0.53
Phase:	DBWAX
Thickness, µm:	1
Temperature limit, °C:	250

Column temperature:	
Initial temperature, °C:	90
Initial hold time, min:	1.0
1 <sup>st</sup> Ramp rate, °C/min:	5
Final temperature, °C:	150
Final hold time, min:	0 (Comment 4)
2 <sup>nd</sup> Ramp rate, °C/min:	30
Final temperature, °C:	220
Final hold time, min:	2.67
Detector:	
Type:	FID
Temperature, °C:	300
H <sub>2</sub> , mL/min:	35
Air, mL/min :	450
N <sub>2</sub> (make-up), mL/min:	40 (Comment 5)
Headspace Sampler	
Type	HP
Model Number	7694
Carrier gas	He (Comment 3)
Pressurization gas	He
Sample Loop, mL:	1.0
Oven, °C:	100
Shake:	Low
Loop Temperature, °C:	110
Transfer Line, °C:	110
GC Cycle Time, min:	20
Vial EQ Time, min:	40 (Comment 6)
Pressurization Time, min:	0.2
Loop Fill Time, min:	0.1
Loop EQ Time, min:	0.05
Inject Time, min:	1.00

B. Calibration

**Preparation of Standard Solutions (Comment 7)**

1. Weigh 0.10 g ( $\pm$  0.1 mg) of ETC into a 100-mL volumetric flask. Add 90 mL of n-hexane. Sonicate until dissolved. Equilibrate to room

temperature. Dilute to volume with n-hexane. Label the flask “1 mg/mL ETC Stock Solution”.

2. Accurately weigh into four sample vials 1.0 g ( $\pm 0.1$ mg) of hexazinone analytical standard.
3. Using a micropipettor, add to the first vial, 10  $\mu$ L of the 1 mg/mL ETC Stock Solution. Label vial as “10 ppm ETC Standard”.
4. Using a micropipettor, add to the second vial, 25  $\mu$ L of the 1 mg/mL ETC Stock Solution. Label vial as “25 ppm ETC Standard”.
5. Using a micropipettor, add to the third vial, 50  $\mu$ L of the 1 mg/mL ETC Stock Solution. Label vial as “50 ppm ETC Standard”.
6. Gently evaporate the n-hexane from the standard vials using a low  $N_2$  purge.
7. After the sample has reached apparent dryness, securely attach the septum and cap to the vial.
8. Equilibrate the gas chromatograph and headspace analyzer.
9. Inject, singly, 1 mL from each standard vial.

**NOTE:** Due to the nature of the sampling system, the standard and sample should have as close to the same weight and chemical make-up as possible. (See Comment 8)

C. Sampling

Sample should be homogeneous and representative of the bulk material from which it was obtained.

D. Sample Analysis

**Preparation of Sample Solutions**

1. Accurately weigh 1.0 g ( $\pm 0.1$  mg) of the sample (in duplicate) into sample vials.
2. Securely attach the septum and cap to each vial.

**Analysis**

1. Equilibrate the gas chromatograph and headspace analyzer.
2. Inject, singly, 1 mL from each sample vial.

E. Calculations

1. Prepare a calibration curve for ETC by plotting the peak area vs. concentration (ppm), corrected for purity, for each standard. (Comment 9).

2. Using the method of least squares, calculate the equation for the best line fit. The correlation coefficient of the least squares equation should be 0.99, or better.
3. Determine the sample impurity concentration (ppm) of the impurity of interest using the equation for the calibration curve,  $y = mx + b$ .

$$\text{Sample result (ppm)} = \frac{\text{peak area} - \text{y intercept}}{\text{slope}}$$

**IX. Quality Control**

1. Obtain a well-characterized sample of test material.
2. Label the sample as "ETC Audit".
3. Analyze the audit sample according to the documented procedure to determine, employing appropriate SQC techniques, if the analysis is on aim.
4. When SQC analysis indicates the analysis is off aim, appropriate troubleshooting and corrective actions should be taken including recalibration of the method/instrument.

**X. Comments**

1. Second laboratory data for inter-laboratory reproducibility was not generated under GLP.
2. Initial method development was performed employing an HP 7694 headspace analyzer. Validation of the method was performed employing a Leap Technologies Combi-PAL autosampler with the following parameters.

Headspace Sampler	
Type:	Leap Technologies
Model Number:	Combi-PAL
Injection Volume, mL:	1
Oven, °C:	100
Incubation Time, min.:	40
Sample Shaker, rpm:	250 - 10 sec. On, 20 sec. off
Syringe Temperature, °C:	130
Syringe Fill Speed, µL/sec:	50
Syringe Injection Time, µL/sec:	750
GC Cycle Time, min:	30

3. Hydrogen may be substituted as carrier gas maintaining constant flow (linear velocity). Hydrogen is flammable and explosive in mixtures with air above 4% and below 75% by volume. It is recommended that hydrogen only be utilized as carrier gas employing GC's equipped with electronic pressure control and safety shutdown devices (e.g., Agilent 6890).
4. Addition of an isothermal hold time at 150°C may be implemented to allow ethyl carbamate to clear the column prior to the final thermal gradient segment.
5. Use of nitrogen makeup gas for the FID is recommended to optimize performance.
6. Equilibration times for the headspace analyzer may need to be lengthened in some cases to obtain reproducible results.
7. Method validation was performed with a 5-point calibration at 10 ppm, 20 ppm, 30 ppm, 40 ppm and 50 ppm. For routine operation the calibration may be reduced to a 3-point calibration curve at 10 ppm, 25 ppm and 50 ppm.
8. Headspace analysis relies on equilibration between the solid, liquid and vapor phase in both the sample and the standard. The standard and sample should have as close to the same weight and chemical make-up as possible. It is recommended for optimal accuracy that calibration be performed utilizing the hexazinone technical sample to be analyzed as the calibration matrix.
9. Peak height may be utilized in place of peak area for this analysis.

**XI. References**

Hewlett Packard HP7694 Headspace Sampler Operating and Service Manual.  
(Hewlett Packard Co., Valley Forge, PA 19842.)

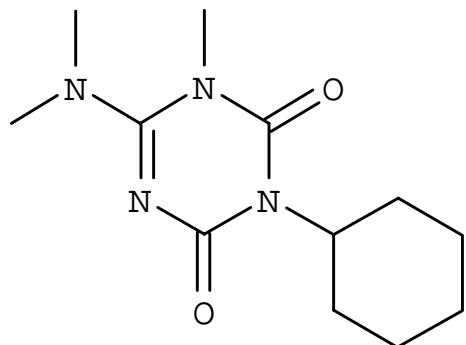
**XII. Appendix**

- A. Chemical Structures
- B. Chromatograms
- C. Calibration Curve

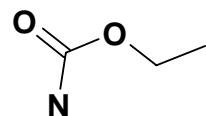
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APPENDIX A

Chemical Structures



Hexazinone



Ethyl Carbamate

## APPENDIX B

### Chromatograms

GC Chromatogram of a Hexazinone Standard Containing 25.0 ppm of Ethyl Carbamate.

Oxygen Research

Gas Chromatography      Ethyl Carbamate Analysis      Data Report  
Sample: 25 ppm std      Dataset 051705 052305  
Misc.: P1397, LS310-1, C72645      062005 Acq On:5/23/05 8:47:16->5/23/05 8:4->  
DPX-A3674-341      (C)

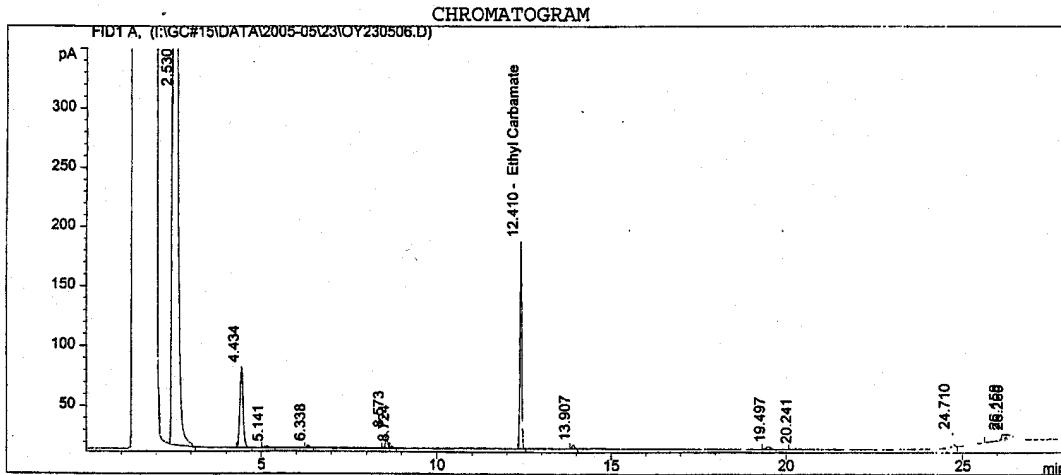
FID1 A,

Inst: PC03043  
Operator: DJarrell  
Raw Data Filename: I:\GC#15\DATA\2005-05\23\OY230506.D  
Aquisition Method: P1397.M  
Analysis Method: I:\GC#15\METHODS\P1135.M

Analysis Report

Retention Time (min)	Compound Name	Peak Width	Peak Area
2.530		0.118	7517.554
4.434		0.096	435.514
5.141		0.108	13.688
6.338		0.079	11.969
8.573		0.063	54.778
8.724		0.067	7.159
12.410	Ethyl Carbamate	0.055	615.433
13.907		0.076	16.977
19.497		0.111	19.810
20.241		0.116	10.344
24.710		0.044	31.873
26.159		0.042	10.398
26.260		0.034	3.385

Total Area  
8748.883



APPENDIX B (Cont.)

UChromatograms

GC Chromatogram of a Hexazinone Sample (DPX-A3674-341) Containing 10 ppm of Ethyl Carbamate.

Oxygen Research

Gas Chromatography      Ethyl Carbamate Analysis      Data Report  
Dataset 051705

Sample: 10 ppm      Acq On: 5/18/05 5:41:17->5/18/05 5:4->  
Misc.: P1397, L5033-1, C68706  
DPX-A3674-341  
LOQ Determination

FID1 A,

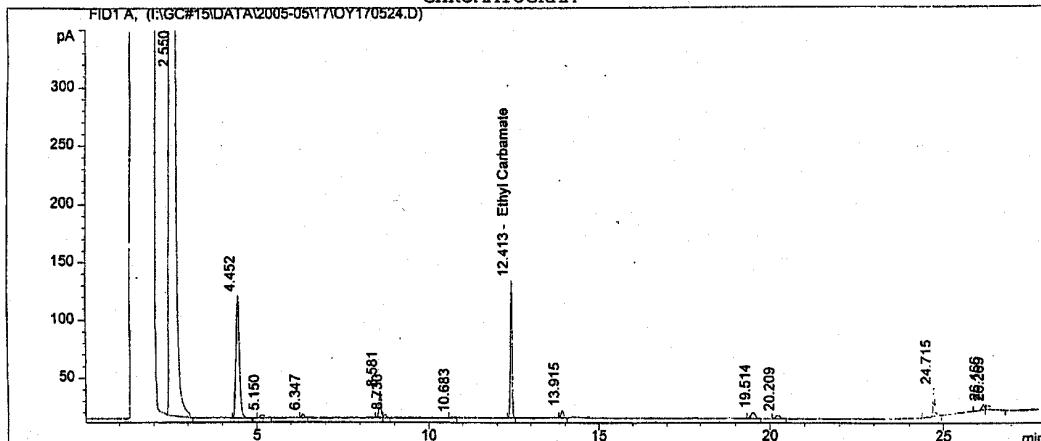
Inst: PC03043  
Operator: DJarrell  
Raw Data Filename: I:\GC#15\DATA\2005-05\17\OY170524.D  
Aquisition Method: P1397.M  
Analysis Method: I:\GC#15\METHODS\P1135.M

Analysis Report

Retention Time (min)	Compound Name	Peak Width	Peak Area
2.550		0.125	11218.566
4.452		0.098	714.254
5.150		0.128	24.085
6.347		0.080	16.606
8.581		0.062	94.140
8.730		0.069	13.776
10.683		0.080	5.297
12.413	Ethyl Carbamate	0.057	422.508
13.915		0.077	30.286
19.514		0.127	43.799
20.209		0.143	25.186
24.715		0.055	83.755
26.166		0.070	17.988
26.269		0.173	47.575

Total Area  
12757.822

CHROMATOGRAM



APPENDIX B (Cont.)

Chromatograms

GC Chromatogram of a Hexazinone Sample (DPX-A3674-341) Unfortified.

Oxygen Research

Gas Chromatography      Ethyl Carbamate Analysis      Data Report  
Dataset 051705  
Sample: Blank      Acq On: 5/17/05 6:23:19 -> 5/17/05 6:2->  
Misc.: P1397, L5033-1, C68706  
DPX-A3674-341

FID1 A,

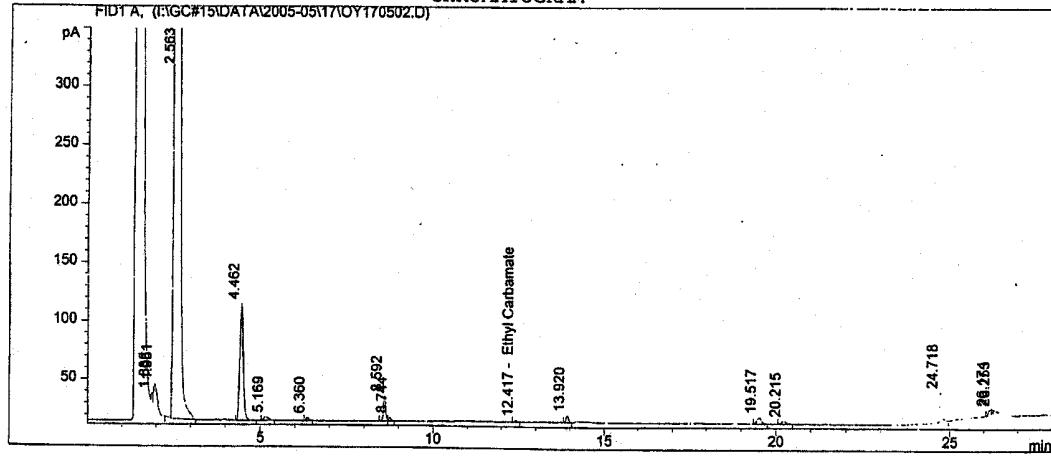
Inst: PC03043  
Operator: DJarrell  
Raw Data Filename: I:\GC#15\DATA\2005-05\17\OY170502.D  
Aquisition Method: P1397.M  
Analysis Method: I:\GC#15\METHODS\P1135.M

Analysis Report

Retention Time (min)	Compound Name	Peak Width	Peak Area
1.885		0.050	22.041
1.951		0.086	94.203
2.563		0.122	10667.997
4.462		0.101	663.104
5.169		0.110	25.014
6.360		0.083	14.599
8.592		0.064	87.717
8.744		0.066	13.260
12.417	Ethyl Carbamate	0.057	4.954
13.920		0.077	28.704
19.517		0.134	44.841
20.215		0.139	23.109
24.718		0.059	97.477
26.174		0.081	17.342
26.253		0.026	2.894

Total Area  
11807.255

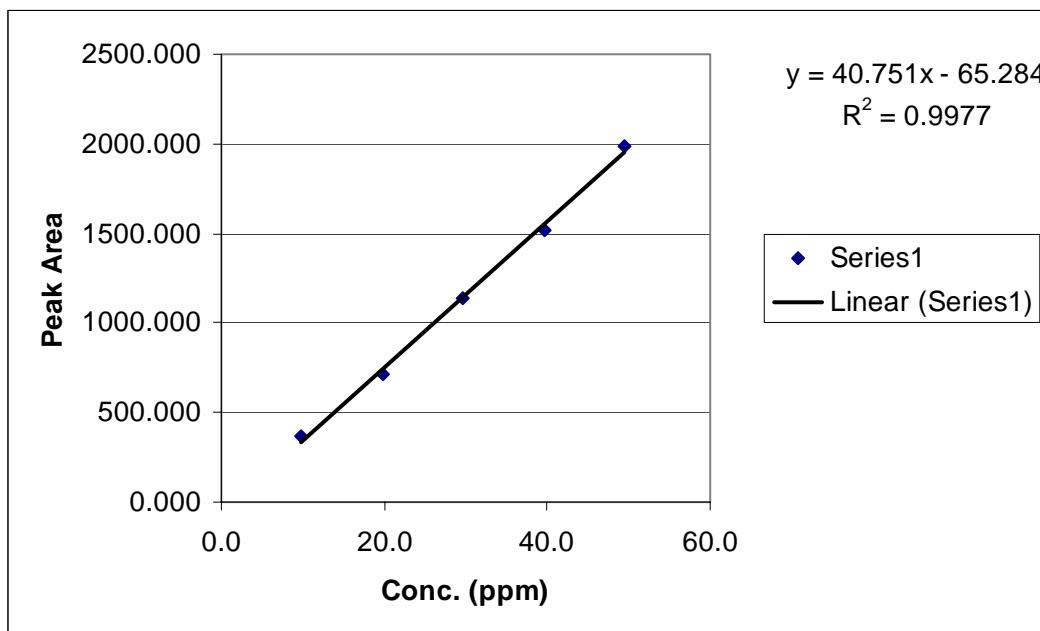
CHROMATOGRAM



APPENDIX C

Calibration Curve

Representative calibration curve (5-point) for determining ethyl carbamate obtained by plotting the ethyl carbamate peak area vs. the concentration of ethyl carbamate. Slope = 40.751; y-intercept = 65.284; Correlation coefficient ( $r$ ) = 0.9988.





## 4. Appendices

### Appendix 1 Analytical Method of Hexazinone TGAI for Ethyl Carbamate (NO. NC2009038C

#### 1. Introduction

Test Item:	Hexazinone TGAI
Classification:	Herbicide
Active Ingredient:	Hexazinone
Empirical Formula:	C <sub>12</sub> H <sub>20</sub> N <sub>4</sub> O <sub>2</sub>
Molecular Weight:	252.3
CAS Register Number:	51235-04-2
IUPAC Name:	3-cyclohexyl-6-dimethylamino-1-methyl-1,3,5-triazine-2,4-(1H,3H)-dione
Structure Formula:	

Impurity Information		
Impurity Name	Empirical Formula	Structure



Impurity I	$C_3H_7NO_2$	
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## 2. Summary

Limit Test of Impurity I in Hexazinone TGAI is performed through HPLC-MSD system, separated on a ZORBAX XDB C8 reverse phase column, detected by MS detector in SIM mode with extract ion of 90 and 112, and analyzed by comparing with external standard.

## 3. Determination of Impurity I in Hexazinone TGAI

### 3.1 Apparatus

Test system: HPLC-MSD 1100 system (Agilent Company).  
Glassware: Volumetric;  
Balance: Mettler Toledo XS 205;  
Vials: 2 ml volume.

### 3.2 Reagents

Impurity I standard: known purity;

Hexazinone TGAI for analysis:

Acetic acid: Analytical grade;  
Acetonitrile: HPLC grade;  
Water: Redistilled;

## 4. Experimental Conditions for Chromatography



The following conditions have been established using an Agilent 1100 HPLC/MSD system. Chromatographic conditions may be changed to obtain satisfactory performance with other instruments provided adequate resolution and sensitivity are achieved.

Equipment: Agilent HPLC-MSD 1100 system.

Column: ZORBAX C8 150 x 4.6 mm, 5 micron

Mobile phase:

Time (min)	Acetonitrile (%)	Acetic acid aqueous solution (pH = 2.8 - 3.2)
0	10	90
3	10	90
10	90	10

Column Temperature: 30°C

Sample size injected: 5.0 µl

Stop time: 10.0 min

Post time: 5 min

Drying gas flow N<sub>2</sub>: 11.0 L/min,

Drying Gas Temperature: 350 °C

Ionization Mode: API-ES

Polarity: Positive

Neb Press: 45 psig

Capillary voltage: 4000 V

Fragmentor: 70 V for SIM run

Gain EMV: 1.0

Actual Dwell: 289

SIM ions: 90 and 112 m/Z

Retention Time: Impurity I: ca 3.0 min

## **5 Limit test of Impurity I**

### **5.1 Preparation of stock Impurity I standard solutions**



Accurately weigh about 10 mg analytical standard grade Impurity I into a 100 ml volumetric flask, dissolve and make to volume with acetonitrile.

#### **5.2 Preparation of Impurity I LOQ standard solution**

Prepare the Impurity I standard solution at acceptable LOQ concentration obtained from the method validation of LOQ by diluting certain amount stock Impurity I standard solution 5.1 with acetonitrile.

#### **5.3 Preparation of sample solution**

Prepare in duplicate per batch.

Approximately 100 mg of Hexazinone TGAI is accurately weighed into a 10 mL volumetric flask, dissolved and made to volume with acetonitrile. The flask is mixed thoroughly.

Another approximately 100 mg of Hexazinone TGAI with same batch is accurately weighed into a 10 mL volumetric flask, dissolved and made to volume with Impurity I LOQ standard solution 5.2. The flask is mixed thoroughly.

#### **5.4 Preparation of blank solution**

An acetonitrile blank solvent solution is prepared.

#### **5.5 Analysis**

When the instrument gets to equilibrium, inject the blank solution. Then, inject Impurity I LOQ standard solution and sample solutions in the following sequence: Impurity I LOQ standard solution, twice non-spiked sample solution, twice spiked sample solution, Impurity I LOQ standard solution. Determine the area of the Impurity I peak.



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### 5.6 Result

Compare Peak Area of Impurity I with LOQ standard solution. The corrected area in spiked samples is calculated as the subtraction of peak area in the spiked sample with the average peak area of the 2 nearest injections of the impurity I LOQ standard solution.

If both the area in TGAI and the corrected area in spiked sample are less than the area of LOQ standard solution, the content of Impurity I in the Hexazinone TGAI is given a limit test result of < LOQ (equivalent to W/W in Test Item). Otherwise, the quantitation method for Impurity I will be developed.