



Food and Agriculture
Organization of the
United Nations

13-15 JUNE, 2023

Cadmium content in fertilizers and its impacts on soil health

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INFA coordinator



1st Meeting
of the International
Network on Soil
Pollution (INSOP)



We are dependent on fertilizers; prices have increased and are expected to do so for another two years



Uneven distribution of nutrient stocks

90% of phosphorus (P) reserves are concentrated in five countries, while over half of the potassium (K) is found in three countries.



Social factors change the global outlook

The high concentration of minerals, coupled with the war conflicts, leading to an acute energy and fertilizer crisis.



Supply shortages triggering price hikes and risking farmers

Sanctions and supply cuts in Russia and Belarus are leaving about **40% of the world's potassium supply out of the market**, leading to shortages and rising prices. Energy plays an important role. **Reduced natural gas restricts the fertilizer industry**, leading to supply shortages, especially of nitrogen.



Higher prices leading to soil nutrient crisis

According to the Commodity Price Outlook report, fertilizer prices were 70% higher in 2022 than in 2021, and the World Bank forecasts that high fertilizer prices could remain in place for up to two more years, putting farmers at risk.

Additional challenges in P-fertilizers

Approximately half of the world's agricultural lands are low in P



We are mining

To grow our crops, we have dug up about 22×10^{12} g P from geological phosphate reserves to produce fertilizers.

We have been doing that every year for more than 50 years, quadrupling the environmental flow of P.

The P used in fertilizers is obtained from rock phosphate, which is a non-renewable resource that should be efficiently used



P reserves are concentrated
The phosphate rock we need to produce fertilizers are limited and concentrated in less than 10 countries



There is no replacement

Unlike nitrogen, phosphorus cannot be pulled from the air and, unlike the carbon in our energy system, there is no known replacement



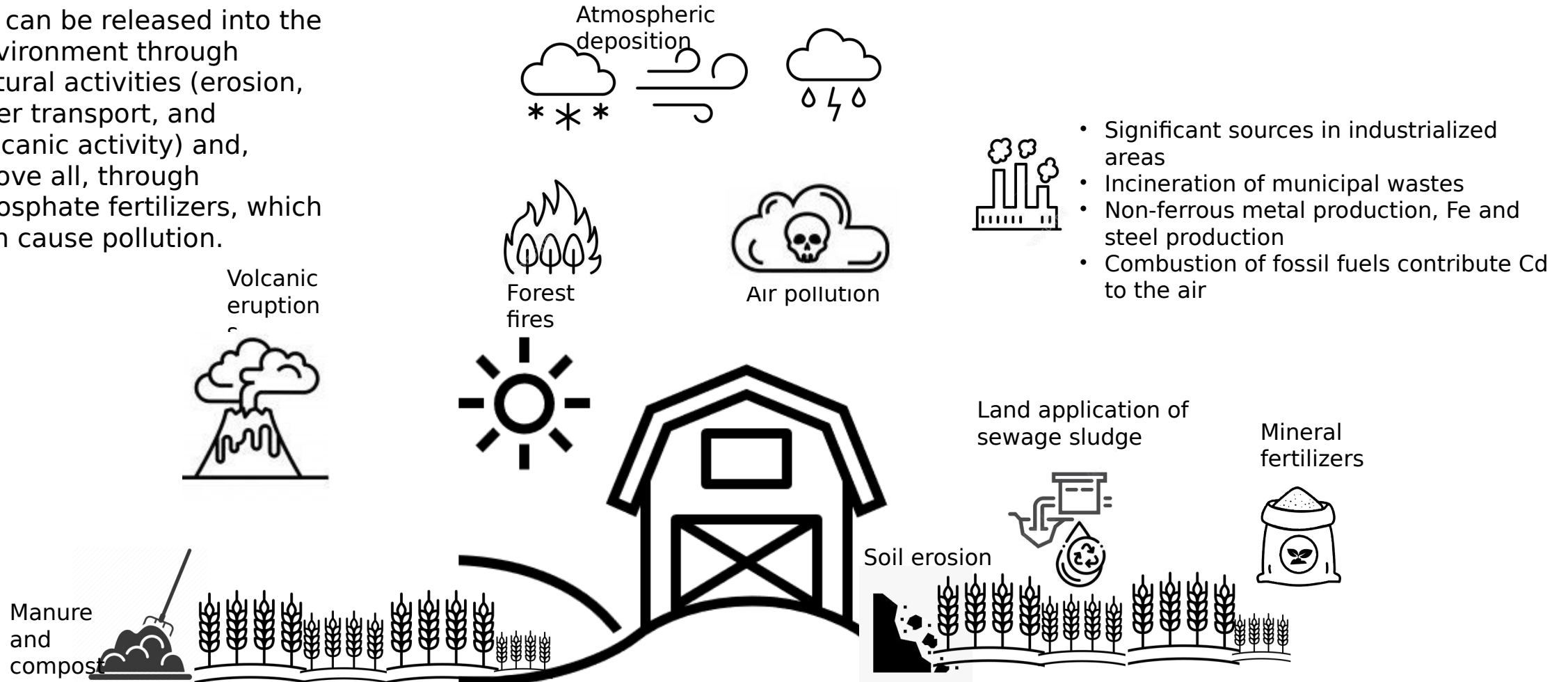
We are wasting P

80 % of phosphorus mined in 2005 was used in fertilizer, but only 17 % made it to the fork (or chopstick). The largest loss — around 46 %— was directly from farms through soil leaching and erosion.

30–40% of food produced is spoiled or wasted, and this wastes around 1 million t P/yr

Where does Cd come from?

Cd can be released into the environment through natural activities (erosion, river transport, and volcanic activity) and, above all, through phosphate fertilizers, which can cause pollution.



Average concentrations of Cd in non-polluted soils are between 0.06 to 1.1 mg/g, with a minimum of 0.01 and a maximum of 2.7 mg/g (Kabata-Pendias and Pendias, 1992)

Quality issue: potential content of heavy metals

Cd persistence in the environment and its uptake and accumulation in the food chain make Cd a public health concern.

- Cadmium (Cd) is a relatively rare metal in nature and belongs to the group of "heavy metals"
- Cd is the third most dangerous pollutant to the environment after mercury (Hg) and lead (Pb).
- It is biopersistent and, once absorbed by an organism, can remain in it for many years.
- The United States Environmental Protection Agency (US EPA, 2021) and the International Agency for Research on Cancer (IARC2021) classify Cd as a human carcinogen due to its ability to accumulate in the body and cause serious and permanent damage.

Table 18.7

SOURCES AND HEALTH EFFECTS OF SELECTED INORGANIC SOIL POLLUTANTS

Chemical	Major uses and sources of soil contamination	Organisms principally harmed ^a	Human health effects
Arsenic	Pesticides, plant desiccants, animal feed additives, coal and petroleum, mine tailings, detergents, and irrigation water	H, A, F, B	Cumulative poison, cancer, skin lesions
Cadmium	Electroplating, pigments, plastic stabilizers, batteries, and phosphate fertilizers	H, A, F, B, P	Heart and kidney disease, bone embrittlement
Chromium	Stainless steel, chrome-plated metals, pigments, refractory brick manufacture, and leather tanning	H, A, F, B	Mutagenic; also essential nutrient
Copper	Mine tailings, fly ash, fertilizers, windblown copper-containing dust, and water pipes	F, P	Rare; mental problems, fatigue; essential nutrient
Lead	Combustion of oil, gasoline, and coal; lead-acid batteries; iron and steel production; solder in water-pipes; paint pigments	H, A, F, B	Brain damage, convulsions
Mercury	Pesticides, catalysts for synthetic polymers, metallurgy, and thermometers; from coal burning	H, A, F, B	Nerve damage
Nickel	Combustion of coal, gasoline, and oil; alloy manufacture; electroplating; batteries; and mining	F, P	Lung cancer
Selenium	High Se geological formations and irrigation wastewater in which Se is concentrated	H, A, F, B	Deformities; essential nutrient
Zinc	Galvanized iron and steel, alloys, batteries, brass, rubber manufacture, mining, and old tires	F, P	Rare; essential nutrient

^aH = humans, A = animals, F = fish, B = birds, P = plants.

Why do fertilizers contain Cd?

Origin	Cadmium content (mg per kg P2O5)		
IGNEOUS	(1)	(2)	(3)
Kola (Russia)	< 13	0.3	0.25
Phalaborwa (South Africa)	< 13	0.1	0.38
SEDIMENTARY			
Florida (USA)	23	19.8 - 32.7	24
Jordan	< 30	12.01.28	18
Khouribga (Morocco)	46	17 - 63	55
Syria	52	13 - 46	22
Algeria	60	42 - 62.6	
Egypt	74		
Bu-Cra (Morocco)	100	101 - 115	97
Nahal Zin (Israel)	100	81 - 112	61
Youssoufia (Morocco)	121	164.7	120
Gafsa (Tunisia)	137	94	173
Togo	162	164 - 179	147
North Carolina (USA)	166	125	120
Taiba (Senegal)	203	165 - 180.6	221
Nauru	243		

Compared with non-phosphate containing rock, sedimentary phosphate rock deposits are about 69 times more enriched with Cd.

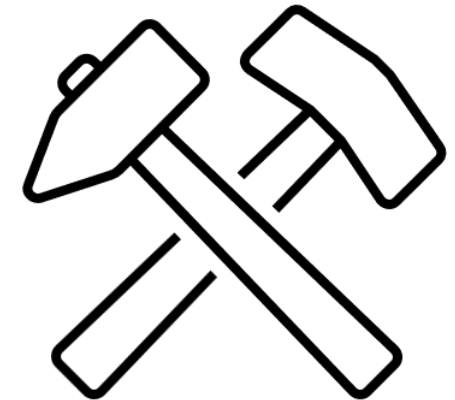
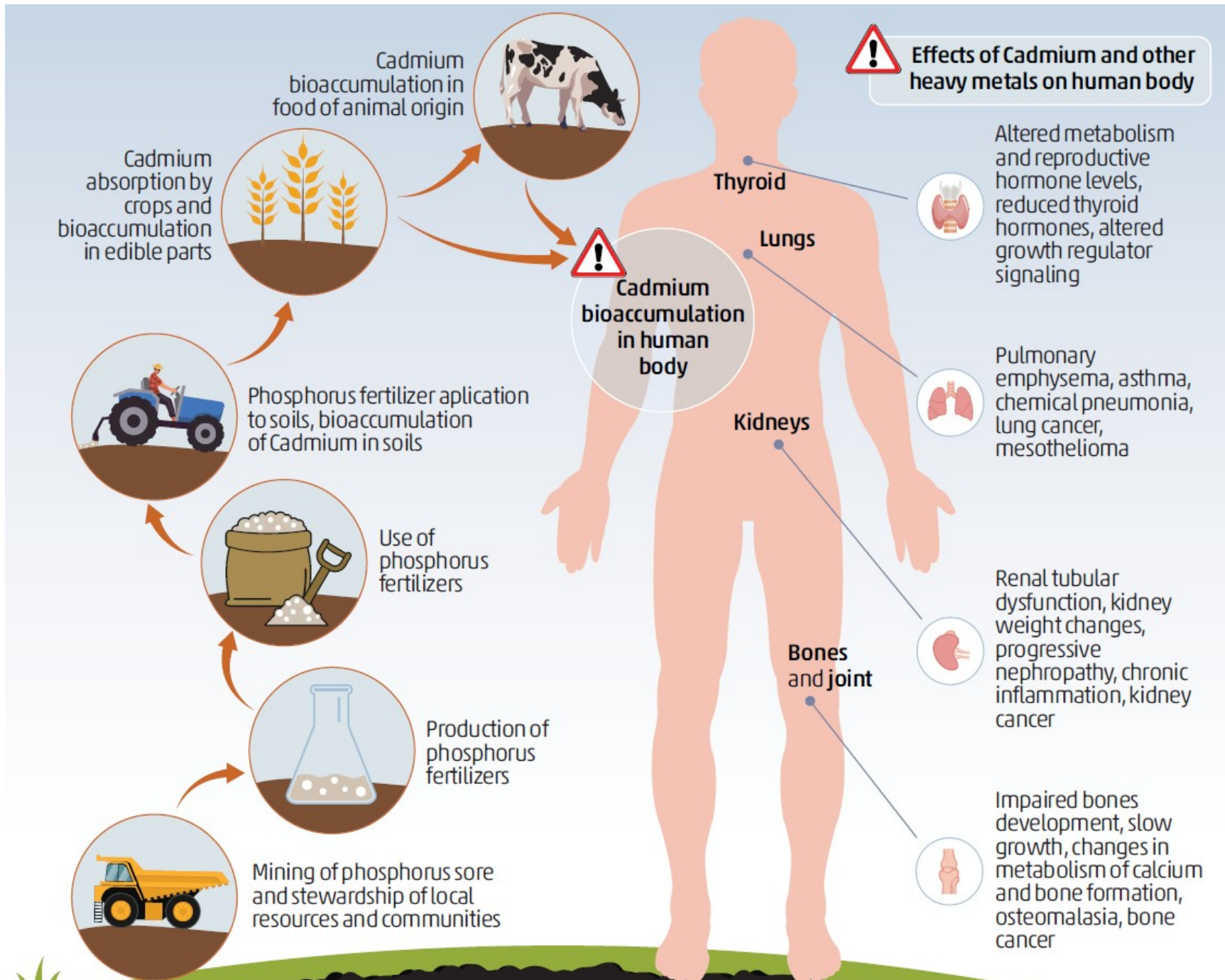


Table: Cadmium contents of main commercial P rocks according to different sources

Sources: **(1)** Davister (1996); **(2)** Botschek and Van Balken (1999); **(3)** Demandt (1999).

The effects of cadmium on health and the environment

The main effect of Cd on human health is kidney disease, and although other adverse effects have been reported (e.g. pulmonary, cardiovascular, and musculoskeletal systems)



Sources and methods for analyzing Cd content

SOURCE	Manure and	Mineral fertilizers	Sampling, sample handling and sample preparation play a significant role
FRACTION	The accuracy of obtained result is strongly related to chemist's decision in choosing the proper and most suitable analysis method. So, the question is, what is the most suitable method?		
EXTRACTION METHODS	The different methods for recovering metals from fertilizers produce highly variable results for a same sample, making comparisons difficult.		
CONCENTRATION DETERMINATION			

Atomic absorption
spectrometry (AAS)

How can INFA and INSOP contribute to monitoring and reducing the contamination and health problems associated to certain sources of mineral P-fertilizers?

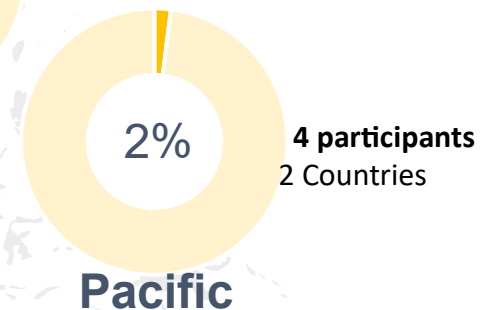
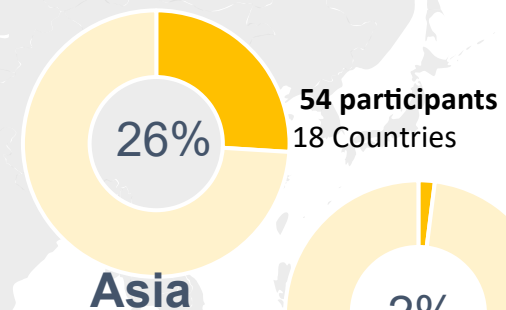
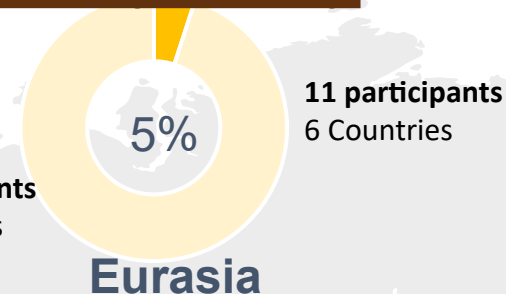
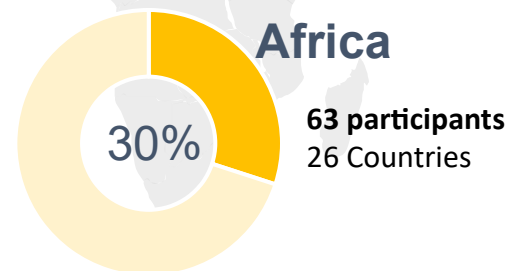
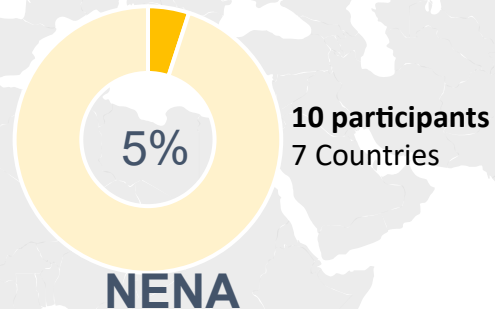
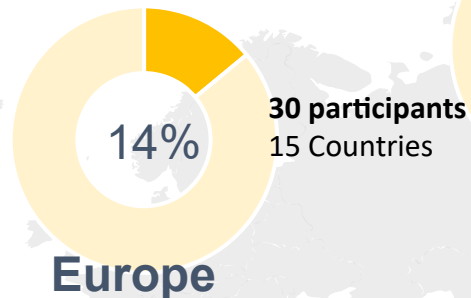
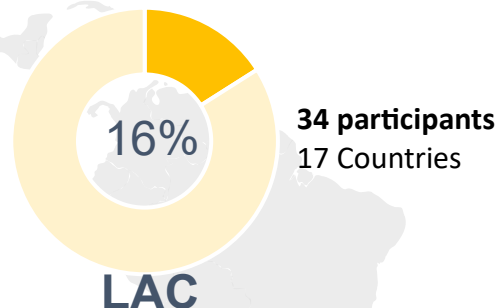
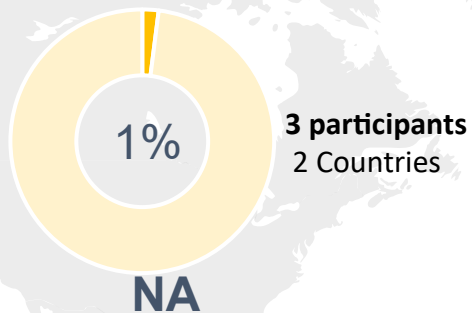


First INSOP-INFA meeting on fertilizer quality assessment

28th February 2023

209 participants
joined virtually to the
INSOP-INFA meeting
on fertilizer quality
assessment

51 % INSOP
49 % INFA



1st Meeting of the International Network on Soil Pollution (INSOP) | 13-15 JUNE, 2023



The proposed areas of collaboration are

A joint collaboration with INFA will address important gaps and, in 2023 will work together towards achieving the following tasks:

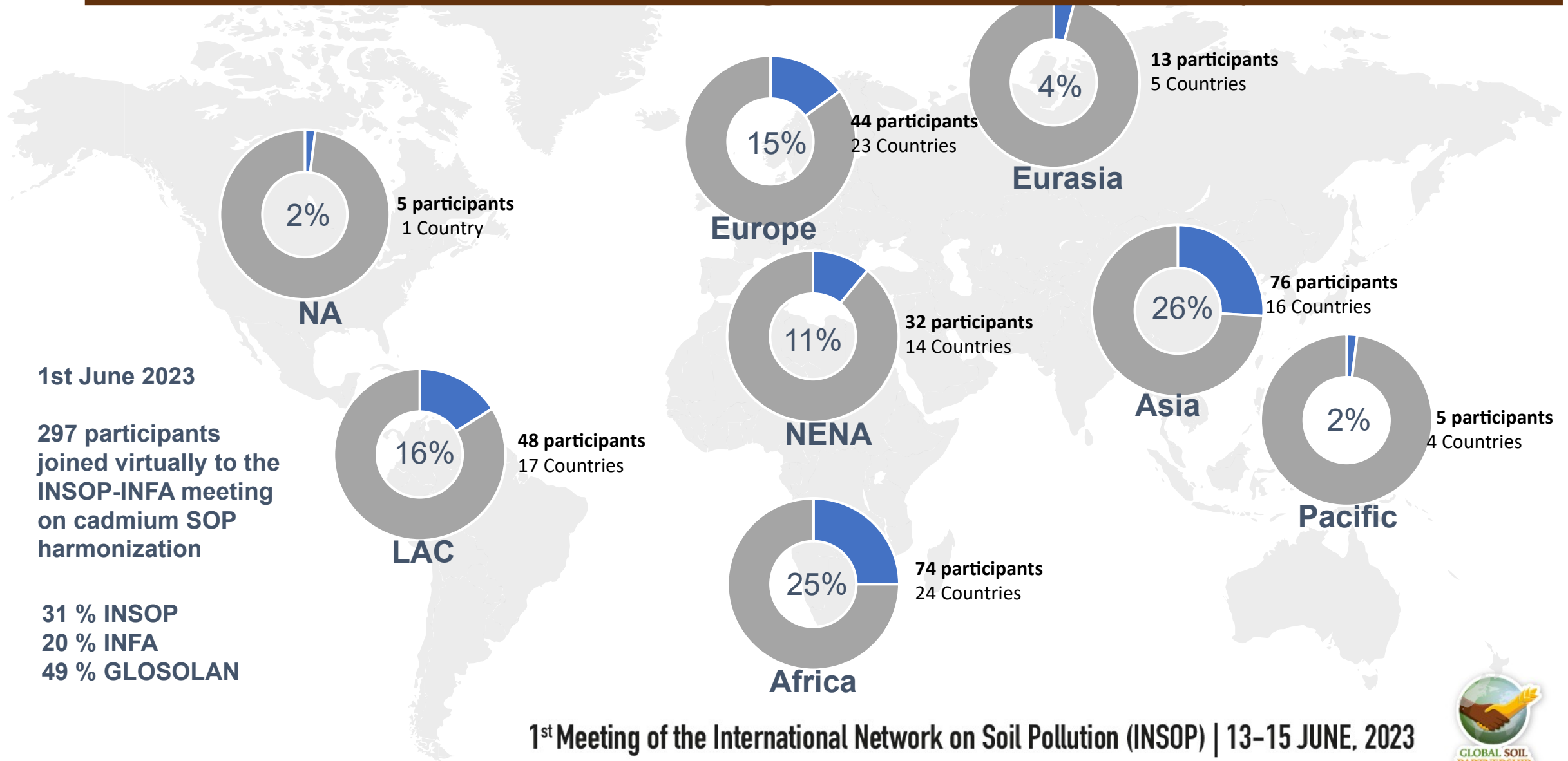
- Identifying the **main fertilizer** sources to be analyzed
- Identifying the most frequently used **methods**
- **Optimizing and harmonizing** of standard operating procedure (SOP) for the analytical evaluation of **heavy metal content in phosphate fertilizers**
- **Raising awareness** on the importance of environmental and health safety of mineral and organic fertilizers and recycled nutrient



Meeting of the International Network on Soil Pollution (INSOP) | 13-15 JUNE, 2023



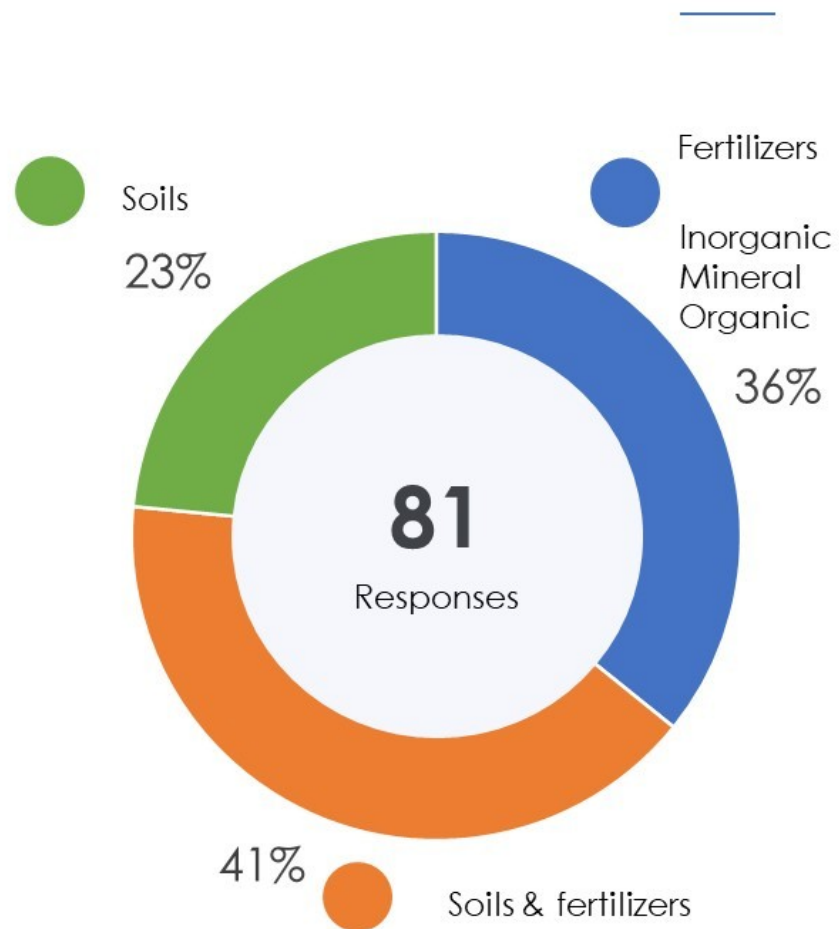
Second INSOP-INFA meeting on fertilizer quality assessment



69 laboratories from 49 countries completed the template for collating information on the methods for Cd content analysis in soils and fertilizers

B	C	D	E	F	G	H	I	J	K	L	M	N	O	P	Q	R	S	T	U
Latin America and the Caribbean	Country	Fertilizer name or type	Fertilizer form	Sample preparation procedure	Sieve size for analysis (mm)	Sample weight (g)	Equipment used for analysis	Reagents Used	Chemical Manufacturers for Reagents	Catalyst is added	Catalyst weight (g)	Volume and concentration of reagents (ml)	Reagent weight (g)	Mixing/Stand time (min)	Analysis type (Digestion, distillation, etc.)	Analysis time (min)	Cost of analysis (US dollars)	Reference materials used	Quality assurance
Pacific	Australia	Compost and Soils	Any form	Dry 105°C	then ring	0.4	ICPMS	Aqua regia (3HCL;1HNO3)	approved	None	NA	20ml			Digestion	120min	element	variety	ISO
Europe and Central Asia	Austria	Soils	solid	Air drying for samples sieved through 2 mm sieve	2 mm	0.5	ICPDES, ICPMS	Aqua regia (6 ml HCL; 2ml HNO3)	VWR	No	NA	6 ml HCL (34-36 %); 2 ml HNO3 (67-69 %)			microwave digestion, ICPDES, ICPMS	Digestion (70 min). Measurement (about 3-6 min per sample)	Prüfstelle		ISO
Asia	Bangladesh	Inorganic and Organic Fertilizer	solid	Drying at 105°C C and Grinding	2mm	2g	AAS	Nitric Acid	BDH, Merck, Fisher or other equivalent manufacturers	No	N/A	10 ml/per sample and 68% HNO3	N/A	30 minutes	Digestion, Volume and filter and then taking the reading in AAS.	Digestion will be continued until the clear digestion formed.	Government organization Approx. 3 US dollar and private 6 US dollar.	CRM	ILC with ACR
Latin America and the Caribbean	Brazil	Soil only	solid (2 mm)	drying at 40 °C	2 mm	0.5 g	ICP-OES	HNO3, H2O2, HCl	Scharlau, Sigma			10 ml			microwave digestion		US\$ 10	2711a Montana soil (NIST)	Lab
Africa	Côte d'Ivoire	Soils, plants and fertilizer	Any form	Air drying for samples sieved through 2 mm sieve	2 mm	0.3g	Digester bloc; distillation unit; Spectrophotometer and Atomic absorption	Sulfuric Acid 98%		Mixed 6g (CuSO4.5 H2O) + 94g (K2SO4)	1	3 ml H2SO4 98%			for total Nitrogen; Spectrophotometer for total Phosphorus and Atomic Absorption for Heavy and light elements		It depends on the parameters but an average of 5 USD per sample	Yes	Yes

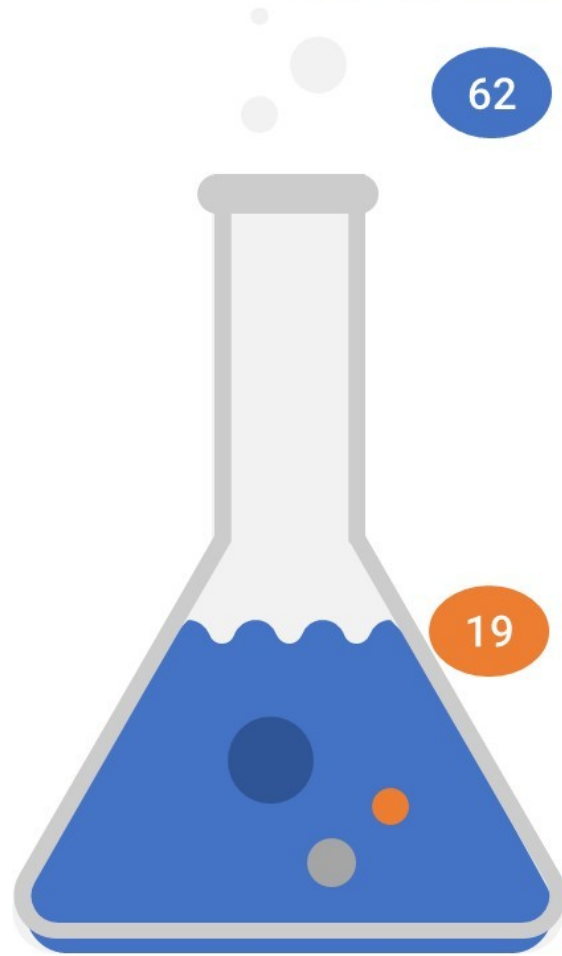
Matrix type



Agreed to:

- Develop and harmonize the SOP on Cd determination in mineral **fertilizers and soils**.
- Mineral fertilizers:
 - multi-nutrient fertilizers (NPK)
 - triple superphosphate (TSP)
 - rock phosphate fertilizers.

Methods for extraction



62

Some form of digestion

Block digestion

Closed digestion

Digestion

Digestion on plate

Hot block extraction and AAS (or MP-AES determination)

Hot block digestion

Microwave

Titration, digestion, and weighting

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Other methods

Dry analysis (no reagents used)

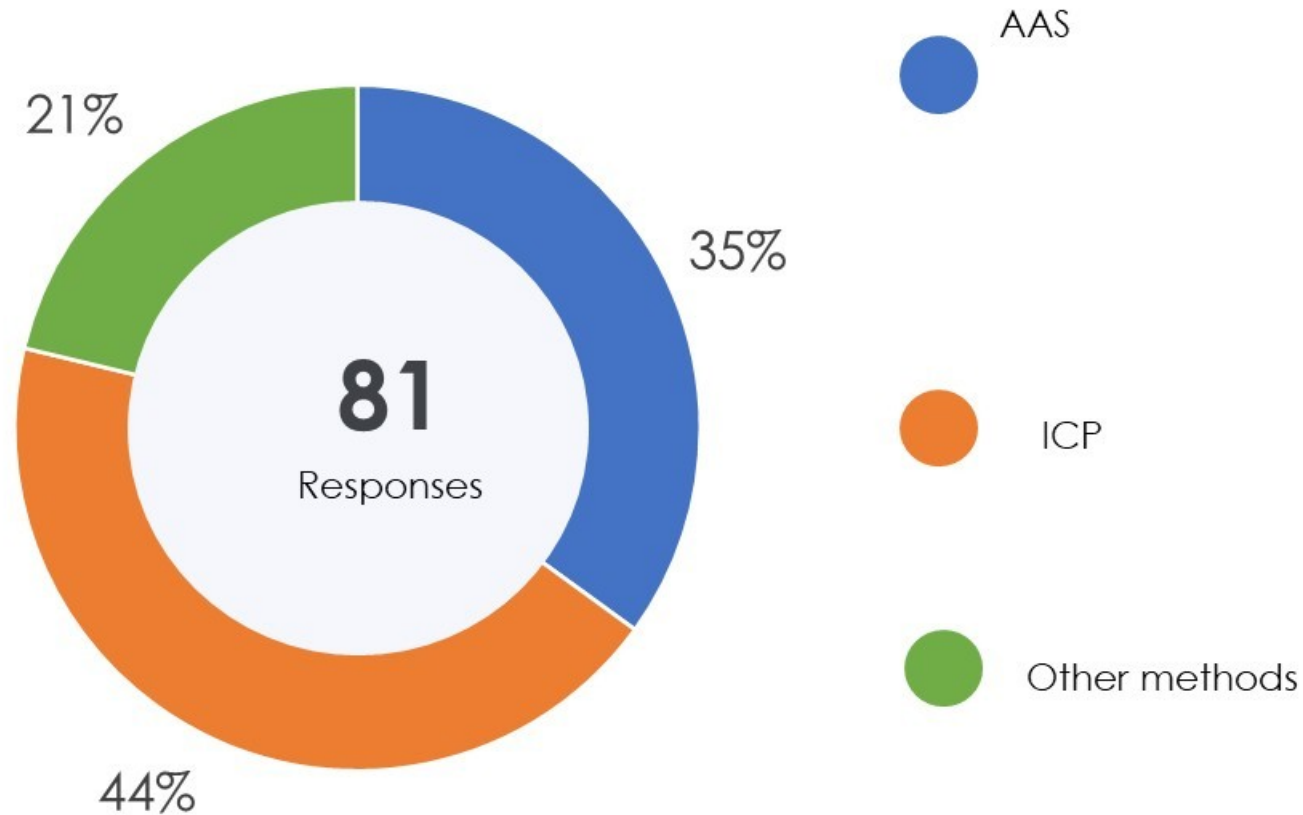
Flux

Filtration

Drying, weighing

- Develop the SOP with two widely used extraction methods: open and microwave digestion.

Equipment for Cd determination



- Develop the SOP with two comprehensive used equipment: ICP and AAS.

Analytical balance
FAAS/GAAS
Graphite atomic absorbance spectroscopy
MPAES
Portable XRF
UV
Flame

1

Methods
selected for
harmonization

Cd content

2

Data sheets developed
to simplify data
capture of method
specifics



3

Data sheets
distributed to and
filled out by INFA-
INSOP-GLOSOLAN
members



5

SOP draft
development and
review
(internal and
external)



4

Harmonization into
centralized
database



6

Publication
and adoption
by labs



Advantages for harmonization

- Provides a standardized and robust procedure to laboratories
 - Procedure works to produce both accurate and precise results
- Ensures results are comparable between laboratories
- Improves fertilizer analysis at a global level
 - Improves resultant decision making
- Capacity building for developing nations

Harmonization process

**INSOP-INFA-GLOSOLAN
teamwork**

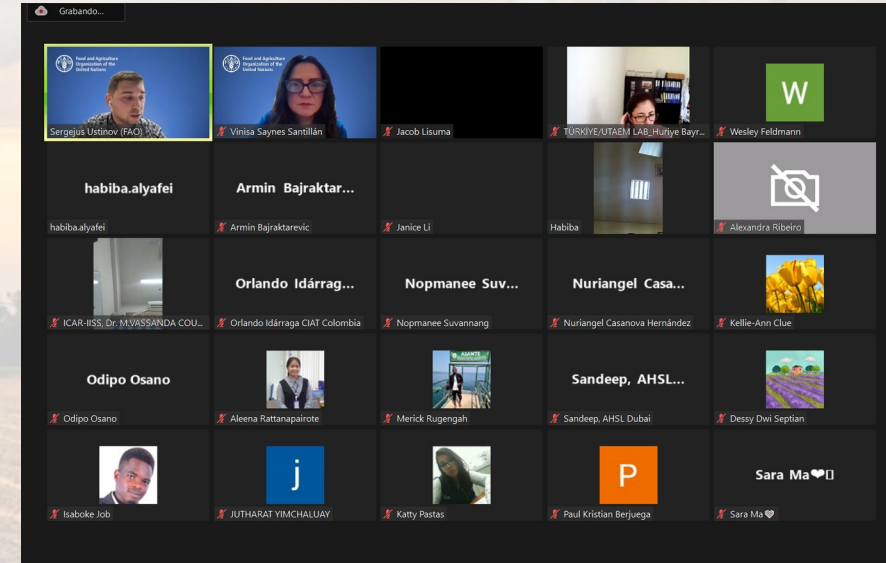
**5 experts agreed to
create a team to help
develop the SOP for
determining Cd levels in
fertilizers and soils**



Thank you for your attention



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Permitted limits in soils and fertilizers

Table 3. Limits for Cd in P fertilizers in several countries expressed as Cd:P ratio, Cd:P₂O₅ or concentration of Cd in the

- Because fertilization increases the risk of Cd transfer to the food chain, some governments have imposed limits
- restricting the Cd content of P fertilizers.

However, it is variable or not defined in different countries

Regardless limits in the different regions **it is necessary to quantify and monitor the heavy metal content with standardized methods**

Austria	75 mg Cd/kg P ₂ O ₅	275	120	54
Belgium	90 mg Cd/kg P ₂ O ₅	206	90	40.5
Denmark		110	48.0	21.6
Netherlands		40	17.5	7.9
Finland	21.5 mg Cd/kg P ₂ O ₅	49	21.5	9.7
Sweden	43 mg Cd/kg P ₂ O ₅	100	43.7	19.7

Cd classification as a human carcinogen goes back to the 1990s. The European Union adopted Regulation (EU) 2019/1009, limiting Cd content in phosphate fertilizers at **60 mg/kg**.