

MONOPOTASSIUM L-GLUTAMATE

Prepared at the 31st JECFA (1987), published in FNP 38 (1988) and in FNP 52 (1992). Metals and arsenic specifications revised at the 57th JECFA (2001). A group ADI 'not specified' for glutamic acid and its Ammonium, Ca, K, Mg & Na salts, was established at the 31st JECFA (1987)

SYNONYMS

Potassium glutamate, MPG; INS No.622

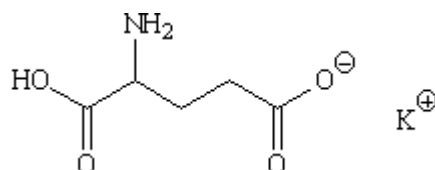
DEFINITION

Chemical names Monopotassium L-glutamate monohydrate,

C.A.S. number 19473-49-5

Chemical formula $C_5H_8KNO_4 \cdot H_2O$

Structural formula



Formula weight 203.24

Assay Not less than 99.0% on the dried basis

DESCRIPTION

FUNCTIONAL USES Flavour enhancer, salt substitute

CHARACTERISTICS

IDENTIFICATION

Solubility (Vol. 4) Freely soluble in water; slightly soluble in ethanol

Test for glutamate (Vol. 4) Passes test

Test for potassium (Vol. 4) Passes test

PURITY

Loss on drying (Vol. 4) Not more than 0.2% (80°, 5 h)

pH (Vol. 4) 6.7 - 7.3 (1 in 50 soln)

Specific rotation (Vol. 4) $[\alpha]_{20, D}$: Between +22.5 and +24.0° (10% (w/v) solution in 2N hydrochloric acid)

Chlorides (Vol. 4) Not more than 0.2%
Test 0.07 g of the sample as directed in the Limit Test using 0.4 ml of 0.01

N hydrochloric acid in the control

Pyrrolidone carboxylic acid Passes test
(Vol. 4)

Lead (Vol. 4)

Not more than 1 mg/kg

Determine using an atomic absorption technique appropriate to the specified level. The selection of sample size and method of sample preparation may be based on the principles of the method described in Volume 4, "Instrumental Methods."

**METHOD OF
ASSAY**

Dissolve about 200 mg of the sample, previously dried and weighed accurately, in 6 ml of formic acid, and add 100 ml of glacial acetic acid. Titrate with 0.1 N perchloric acid determining the end-point potentiometrically. Run a blank determination in the same manner and correct for the blank. Each ml of 0.1 N perchloric acid is equivalent to 10.162 mg of $C_5H_8KNO_4 \cdot H_2O$